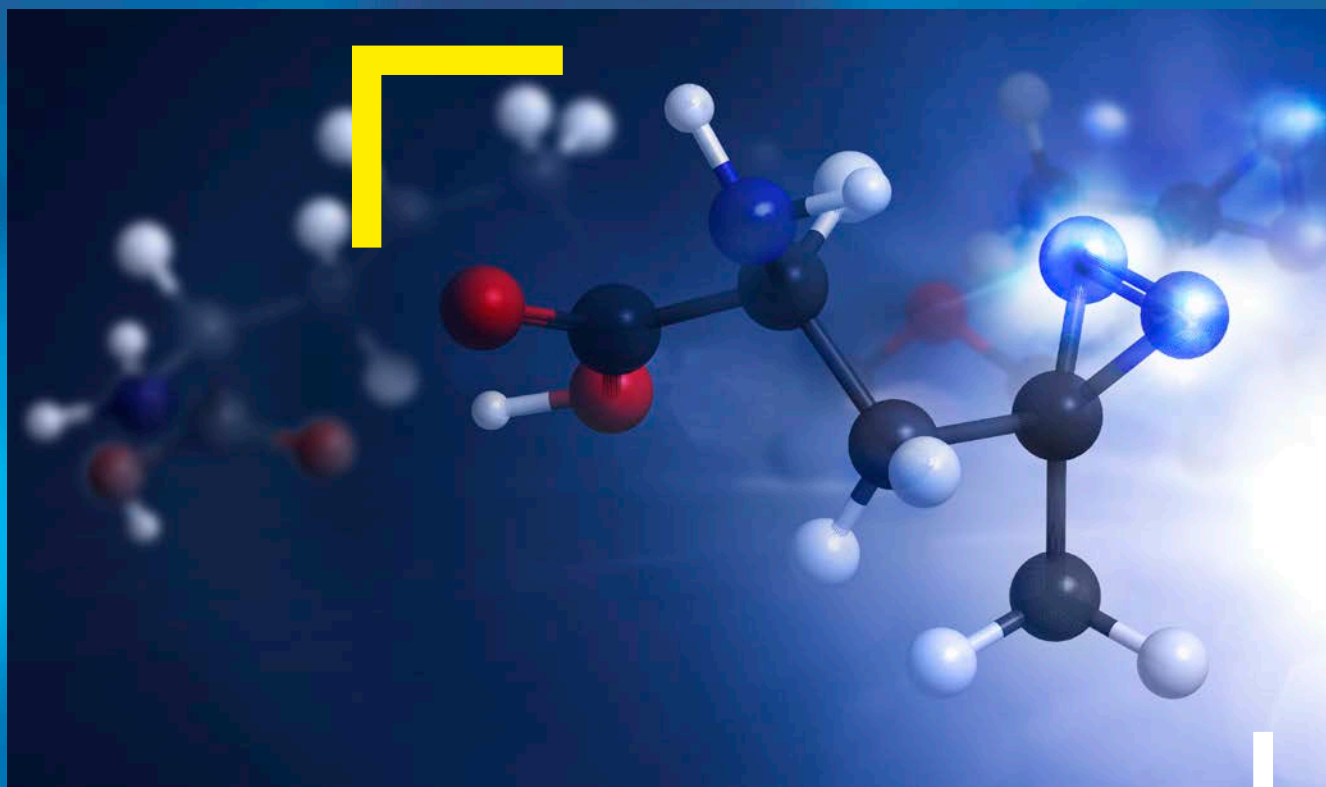


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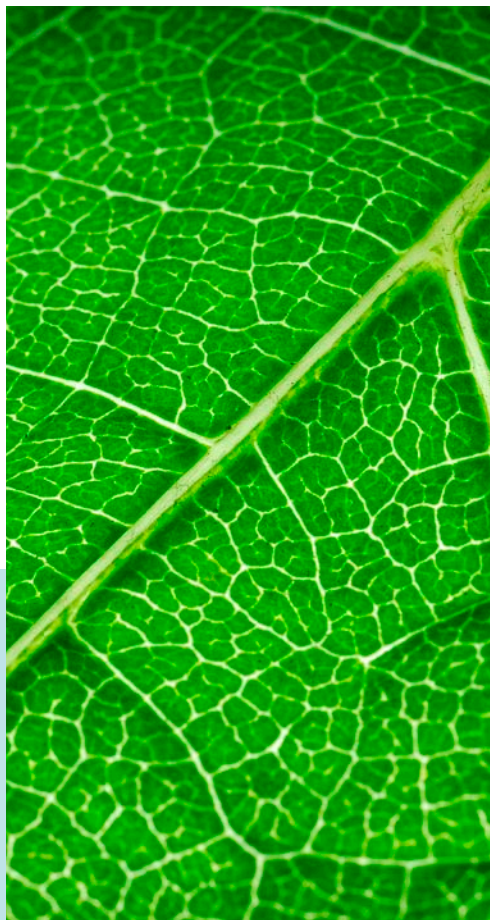
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CHEMISTRY

**PHOTOACTIVATED COMPOUNDS
FOR PEPTIDE SYNTHESIS AND
BIOCONJUGATION**

Empowering Peptide Innovation



Photochemistry in Peptide Synthesis and Bioconjugation

The term photochemistry describes a group of chemical transformations initiated by irradiation with light. Photochemical reactions usually occur at room temperature and normal pressure, and mostly do not require additional reagents or catalysts, with the notable exception of some cases where the presence of a photosensitizer is necessary. Therefore, photochemical transformations are usually orthogonal to classical chemical transformations, a characteristic that renders them a valuable tool for chemists.

Consequently, the applications of photochemistry are numerous. Orthogonality is a trait often sought for in protecting groups and linkers, as it allows for their selective cleavage. Furthermore, photocleavage is a convenient method for selective removal of auxiliaries after their function has been served. The typical moiety that is incorporated into protecting groups, linkers and auxiliaries to facilitate light-induced cleavage is an *o*-nitrobenzyl group, which undergoes a Norrish-type II reaction upon UV-irradiation.

Another common application of photochemistry is the labeling or crosslinking of biomolecules *in vitro* and *in vivo*. The latter is of particular interest as a photochemical reaction is one of the few chemical transformations that can be selectively initiated in living cells.

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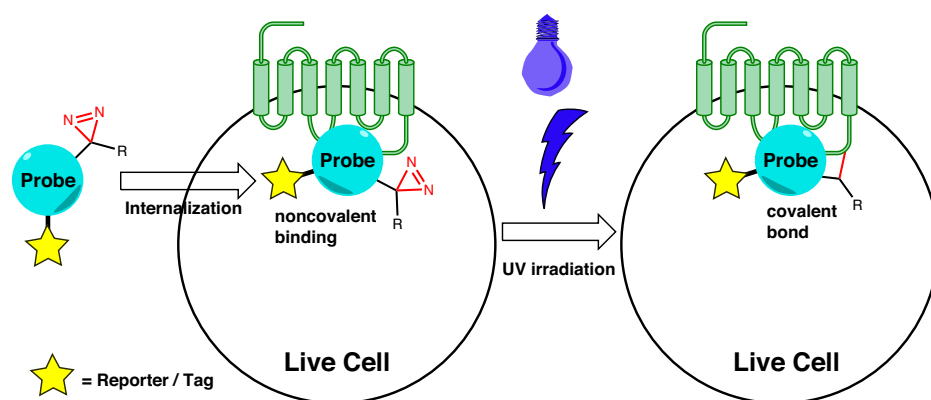
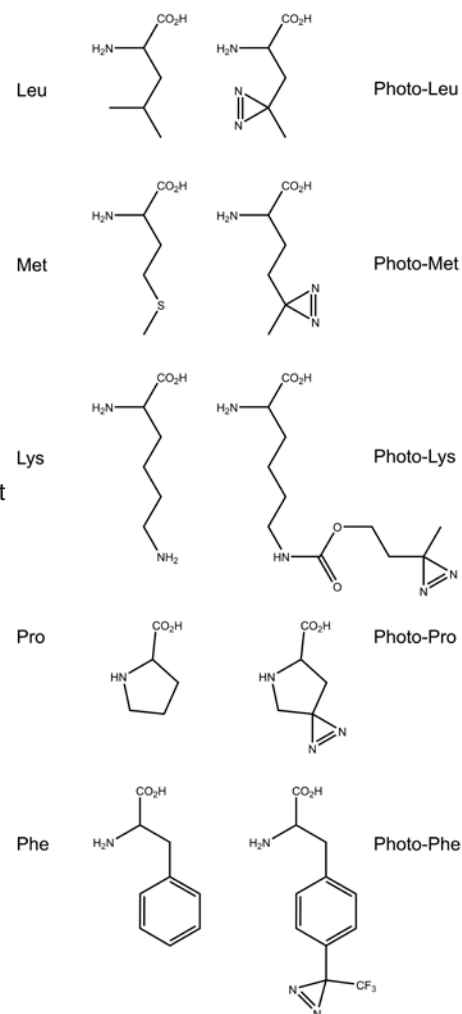
1. Diazirine Amino Acids for Photo-Crosslinkage in Living Cells

Iris Biotech introduces a comprehensive set of photo-crosslinking amino acids bearing the diazirine moiety. Irradiation of diazirines with UV light (ca. 350 nm – 360 nm) yields a highly reactive carbene species that can undergo insertions into C-C, CH, O-H and X-H (X = heteroatom) bonds of neighboring molecules to irreversibly form a covalent bond. The diazirine moiety is the smallest of all photophores, so introduction of a diazirine-bearing amino acid into a peptide or protein usually does not impair its biological activity. Further advantages of diazirine crosslinkers are their stability at room temperature, as well as their relative stability to nucleophiles, and to both acidic and basic conditions.

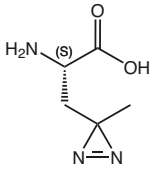
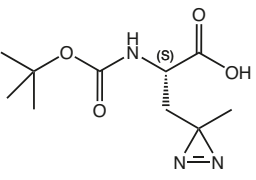
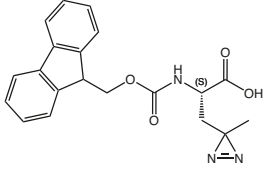
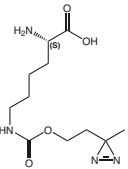
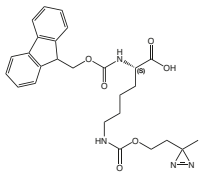
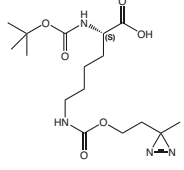
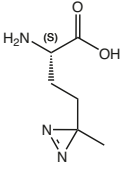
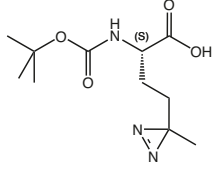
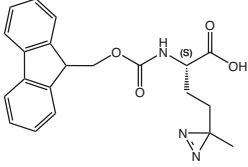
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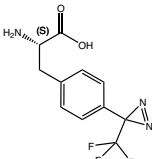
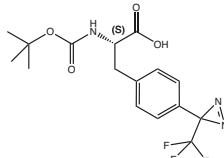
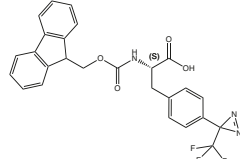
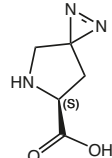
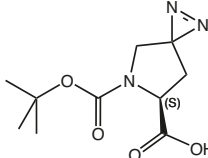
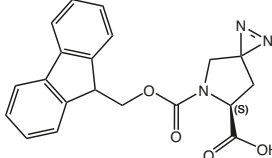
- Probing protein-protein and protein-peptide interactions by photo-crosslinking (e.g. site-specific incorporation of Photo-Lys into glutathione S-transferase allowing covalent crosslinking of the two subunits of the dimeric protein in *E. coli*)
- Probing protein-peptide interactions in order to identify cellular targets of peptides of interest
- Studying protein-drug interactions and identifying new drug targets
- By comparing results obtained from different proteomic setups (e.g. live cells and cell lysates), more putative targets can be identified.

These amino acids are available in Fmoc- as well as Boc-protected versions for their incorporation into synthetic peptides via standard coupling methods. Furthermore, the unprotected diazirine amino acids are also available for incorporation into expressed peptides and proteins by utilizing the appropriate aminoacyl-tRNA synthetase/tRNA pair. A combination of synthetic and recombinant approaches utilizing NCL has been demonstrated as well. Unnatural amino acids are frequently toxic to cells; however, these photo-amino acids are functional and nontoxic, which allows them to be a premium tool for studying mechanisms and interactions in living cells.



Identification of intracellular targets using a diazirine-bearing probe. By attaching a reporter group or tag to the probe, the target of the binding probe can be either identified using the reporter group, or isolated using the tag.

		Article No.	Quantity	Price
HAA3100	H-L-Photo-Leucine*HCl			
<p>(S)-2-amino-3-(3-methyl-3H-diazirin-3-yl)propanoic acid hydrochloride CAS NO: 851960-91-3 FORMULA: C₉H₉N₃O₂*HCl MOLECULAR WEIGHT: 143,14*36,45 g/mol</p>			HAA3100.0250 250 mg € 375,00 HAA3100.1000 1 g € 1.100,00 HAA3100.5000 5 g € 4.250,00	
BAA3070	Boc-L-Photo-Leucine*DCHA			
<p>CAS NO: 1000770-97-7 net FORMULA: C₁₀H₁₇N₃O₄*C₁₂H₂₃N MOLECULAR WEIGHT: 243,26*181,34 g/mol</p>			BAA3070.0100 100 mg € 250,00 BAA3070.0250 250 mg € 590,00 BAA3070.1000 1 g € 1.750,00	
FAA4590	Fmoc-L-Photo-Leucine			
<p>(S)-2-(((9H-fluoren-9-yl)methoxy)carbonylamino)-3-(3-methyl-3H-diazirin-3-yl)propanoic acid CAS NO: 1360651-24-6 FORMULA: C₂₀H₁₉N₃O₄ MOLECULAR WEIGHT: 365,38 g/mol</p>			FAA4590.0100 100 mg € 225,00 FAA4590.0250 250 mg € 490,00 FAA4590.0001 1 g € 1.450,00	
HAA3110	H-L-Photo-Lysine*HCl			
<p>(S)-2-amino-6-((2-(3-methyl-3H-diazirin-3-yl)ethoxy)carbonylamino)hexanoic acid hydrochloride CAS NO: 1253643-88-7 FORMULA: C₁₁H₂₀N₄O₄*HCl MOLECULAR WEIGHT: 272,30*36,45 g/mol</p>			HAA3110.0250 250 mg € 375,00 HAA3110.1000 1 g € 1.100,00 HAA3110.5000 5 g € 4.250,00	
FAA4600	Fmoc-L-Photo-Lysine			
<p>(S)-2-(((9H-fluoren-9-yl)methoxy)carbonylamino)-6-((2-(3-methyl-3H-diazirin-3-yl)ethoxy)carbonylamino)hexanoic acid FORMULA: C₂₈H₃₀N₄O₆ MOLECULAR WEIGHT: 494,54 g/mol</p>			FAA4600.0250 250 mg € 375,00 FAA4600.1000 1 g € 1.100,00 FAA4600.5000 5 g € 4.250,00	
BAA3080	Boc-L-Photo-Lysine			
<p>(S)-2-(tert-butoxycarbonylamino)-6-((2-(3-methyl-3H-diazirin-3-yl)ethoxy)carbonylamino)hexanoic acid CAS NO: 1330088-06-6 FORMULA: C₁₆H₂₈N₄O₆ MOLECULAR WEIGHT: 372,42 g/mol</p>			BAA3080.0250 250 mg € 375,00 BAA3080.1000 1 g € 1.100,00 BAA3080.5000 5 g € 4.250,00	
HAA3120	H-L-Photo-Methionine*HCl			
<p>(S)-2-amino-4-(3-methyl-3H-diazirin-3-yl)butanoic acid hydrochloride CAS NO: 851960-68-4 FORMULA: C₆H₁₁N₃O₂*HCl MOLECULAR WEIGHT: 157,17*36,45 g/mol</p>			HAA3120.0250 250 mg € 375,00 HAA3120.1000 1 g € 1.100,00 HAA3120.5000 5 g € 4.250,00	
BAA3090	Boc-L-Photo-Methionine			
<p>(S)-2-(tert-butoxycarbonylamino)-4-(3-methyl-3H-diazirin-3-yl)butanoic acid CAS NO: 1002754-75-7 FORMULA: C₁₁H₁₉N₃O₄ MOLECULAR WEIGHT: 257,29 g/mol</p>			BAA3090.0250 250 mg € 375,00 BAA3090.1000 1 g € 1.100,00 BAA3090.5000 5 g € 4.250,00	
FAA4610	Fmoc-L-Photo-Methionine			
<p>(S)-2-(((9H-fluoren-9-yl)methoxy)carbonylamino)-4-(3-methyl-3H-diazirin-3-yl)butanoic acid CAS NO: 945859-89-2 FORMULA: C₂₁H₂₁N₃O₄ MOLECULAR WEIGHT: 379,41 g/mol</p>			FAA4610.0250 250 mg € 375,00 FAA4610.1000 1 g € 1.100,00 FAA4610.5000 5 g € 4.250,00	

		Article No.	Quantity	Price
HAA3490 H-L-Photo-Phe-OH 4-(trifluoromethyldiazirin)-L-phenylalanine CAS NO: 92367-16-3 FORMULA: $C_{11}H_{10}F_3N_2O_2$ MOLECULAR WEIGHT: 273,21 g/mol		HAA3490.0050	50 mg	€ 910,00
		HAA3490.0100	100 mg	€ 1.410,00
BAA1530 Boc-L-Photo-Phe-OH N-alpha-(t-Butyloxycarbonyl)-4-(trifluoromethyldiazirin)-L-phenylalanine CAS NO: 133342-64-4 FORMULA: $C_{16}H_{18}F_3N_2O_4$ MOLECULAR WEIGHT: 373,33 g/mol		BAA1530.0100	100 mg	€ 910,00
		BAA1530.0250	250 mg	€ 1.740,00
		BAA1530.0500	500 mg	€ 2.570,00
FAA5690 Fmoc-L-Photo-Phe-OH N-alpha-(9-Fluorenylmethyloxycarbonyl)-4-(trifluoromethyldiazirin)-L-phenylalanine CAS NO: 133342-64-0 FORMULA: $C_{26}H_{20}F_3N_2O_4$ MOLECULAR WEIGHT: 495,45 g/mol		FAA5690.0100	100 mg	€ 740,00
		FAA5690.0250	250 mg	€ 1.410,00
		FAA5690.0500	500 mg	€ 2.070,00
HAA3130 H-L-Photo-Proline*HCl (S)-1,2,5-triazaspiro[2.4]hept-1-ene-6-carboxylic acid CAS NO: 1675206-55-9 FORMULA: $C_5H_7N_3O_2 \cdot HCl$ MOLECULAR WEIGHT: 141,13*36,45 g/mol		HAA3130.0250	250 mg	€ 375,00
		HAA3130.1000	1 g	€ 1.100,00
		HAA3130.5000	5 g	€ 4.250,00
BAA3100 Boc-L-Photo-Proline (S)-5-(tert-butoxycarbonyl)-1,2,5-triazaspiro[2.4]hept-1-ene-6-carboxylic acid CAS NO: 1266778-55-5 FORMULA: $C_{10}H_{15}N_3O_4$ MOLECULAR WEIGHT: 241,24 g/mol		BAA3100.0250	250 mg	€ 375,00
		BAA3100.1000	1 g	€ 1.100,00
		BAA3100.5000	5 g	€ 4.250,00
FAA4620 Fmoc-L-Photo-Proline (S)-5-(((9H-fluoren-9-yl)methoxy)carbonyl)-1,2,5-triazaspiro[2.4]hept-1-ene-6-carboxylic acid CAS NO: 1266778-58-8 FORMULA: $C_{20}H_{17}N_3O_4$ MOLECULAR WEIGHT: 363,37 g/mol		FAA4620.0250	250 mg	€ 375,00
		FAA4620.1000	1 g	€ 1.100,00
		FAA4620.5000	5 g	€ 4.250,00

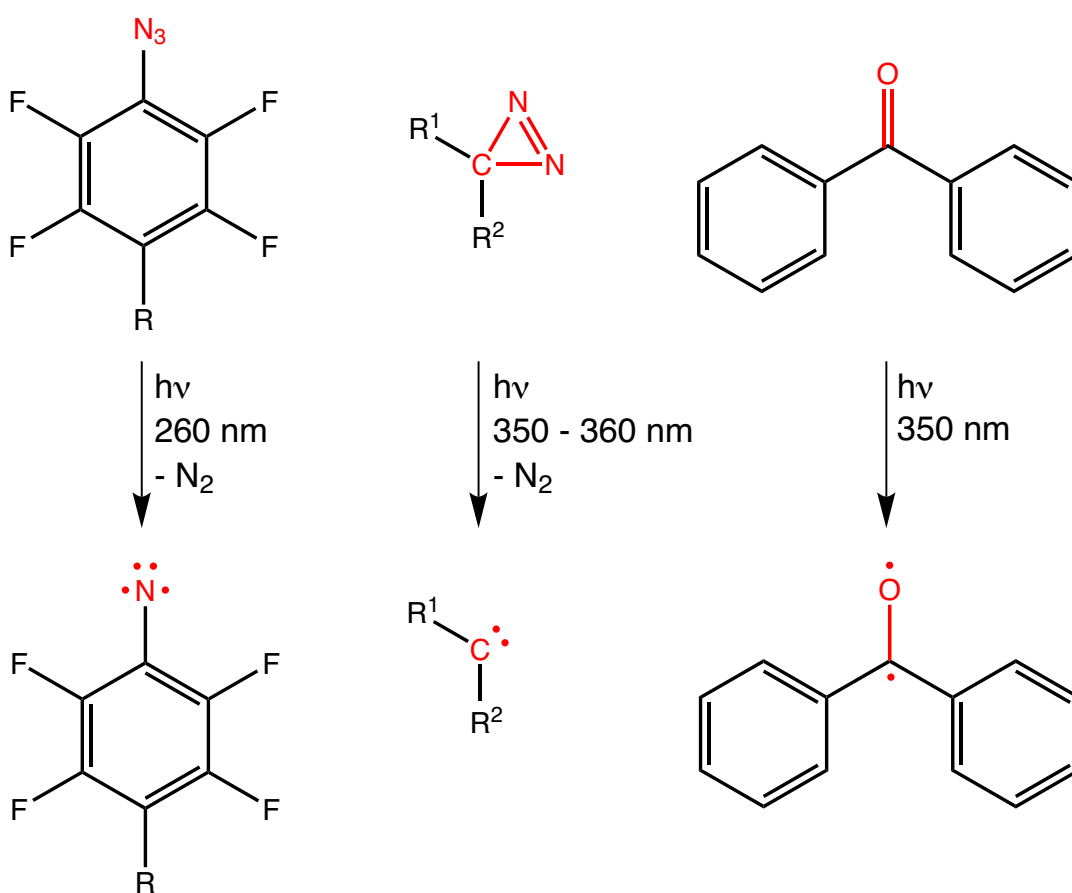
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2. Photo-Crosslinkers for Various Applications

Many photo-crosslinkers for photoaffinity labeling rely on benzophenone as the crosslinking agent. However, these types of crosslinkers usually show low crosslinking yields and require relatively long irradiation times due to slow reaction rates, which may lead to non-specific labeling. Moreover, the irradiation conditions for benzophenones have been shown to lead to cell damage and cell death.

Conversely, diazirine- and perfluorophenyl-based crosslinkers generate a reactive species (carbene and nitrene, respectively) upon relatively short irradiation with UV light. Consequently, diazirine- and perfluorophenyl-crosslinkers are commonly used in molecular biology and biochemistry.

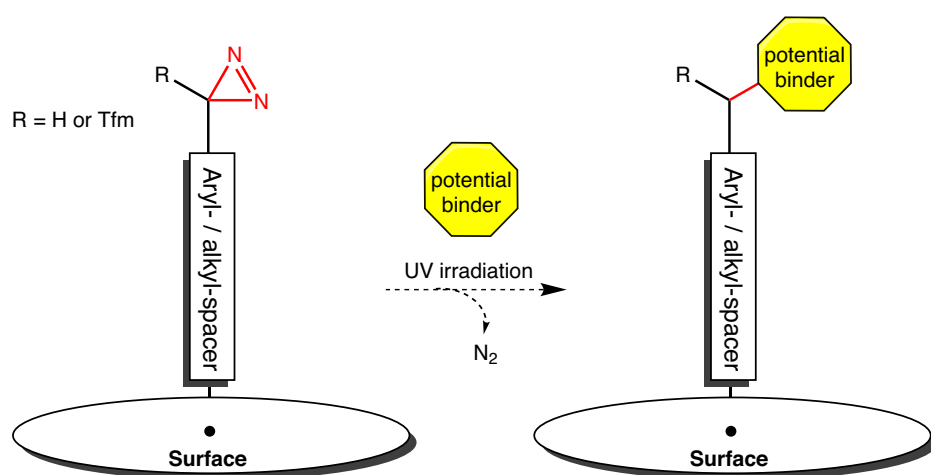


Three types of photophores: Perfluorophenyl azides, diazirines and benzophenones

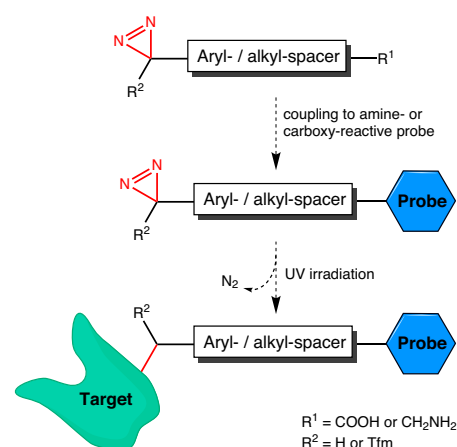
2.1. Photo-Crosslinkers for Various Applications

Diazirine-bearing crosslinkers are activated at wavelengths that cause little to no damage to cells. In addition to this, trifluoromethylaryl diazirines show a fast initial reaction rate and rapid termination of reaction. Consequently, they exhibit a highly ligand-dependent reactivity, which renders them ideal probes for ligand binding to low-affinity targets, e.g. for probing carbohydrate-lectin interactions. Moreover, the small size of the diazirine group minimizes the risk of impairing or altering the biological activity of a ligand. Our diazirine crosslinkers are functionalized to react with either carboxyl- or amine-reactive ligands, respectively.

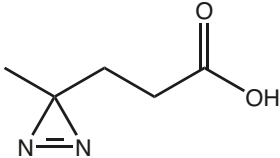
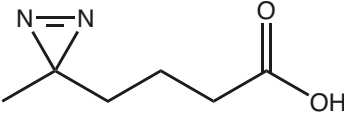
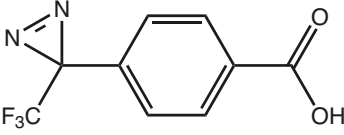
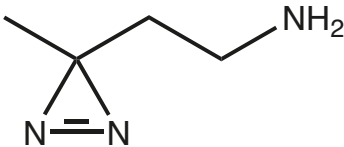
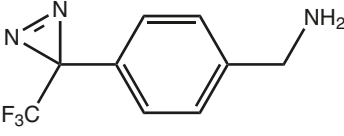
Another application of diazirine photo crosslinkers is the surface immobilization of molecules of interest. The linker is bound to a surface through its carboxyl or amino functionality, leaving the diazirine group free to react with any type of molecule. Since this reaction takes place irrespective of available functional groups, it is not necessary to chemically modify molecules of interest prior to immobilization. This virtually ensures that molecules are immobilized without altering their binding properties. By using this approach, it is possible to easily create microarrays of whole libraries of small molecules for rapid screening.



Immobilization of molecules of interest on a surface (e.g. a glass slide) using a diazirine crosslinker.



Photoaffinity labeling of a target molecule by a diazirine-functionalized binding probe.

		Article No.	Quantity	Price
RL-2890	Photo-Propanoic acid			
<p>3-(3-methyl-3H-diazirin-3-yl)propanoic acid CAS NO: 25055-86-1 FORMULA: C₅H₈N₂O₂ MOLECULAR WEIGHT: 128,13 g/mol</p>				
		RL-2890.0000	please inquire	
RL-2900	Photo-Butyric acid			
<p>4-(3-methyl-3H-diazirin-3-yl)butanoic acid CAS NO: 16297-97-5 FORMULA: C₆H₁₀N₂O₂ MOLECULAR WEIGHT: 142,16 g/mol</p>				
		RL-2900.0000	please inquire	
RL-2920	Photo-Benzoic acid			
<p>4-[3-(Trifluoromethyl)-3H-diazirin-3-yl]benzoic acid CAS NO: 85559-46-2 FORMULA: C₉H₅F₃N₂O₂ MOLECULAR WEIGHT: 230,14 g/mol</p>				
		RL-2920.0200 RL-2920.1000	200 mg 1 g	€ 200,00 € 600,00
RL-2910	Photo-Ethylamine*HCl			
<p>2-(3-methyl-3H-diazirin-3-yl)ethan-1-amine hydrochloride FORMULA: C₄H₉N₃*HCl MOLECULAR WEIGHT: 99,13*36,45 g/mol</p>				
		RL-2910.0000	please inquire	
RL-2930	Photo-Benzylamine*HCl			
<p>4-[3-(Trifluoromethyl)-3H-diazirin-3-yl]benzylamine hydrochloride CAS NO: 1258874-29-1 FORMULA: C₉H₈N₃F₃*HCl MOLECULAR WEIGHT: 215,18*36,45 g/mol</p>				
		RL-2930.0200 RL-2930.1000	200 mg 1 g	€ 250,00 € 700,00

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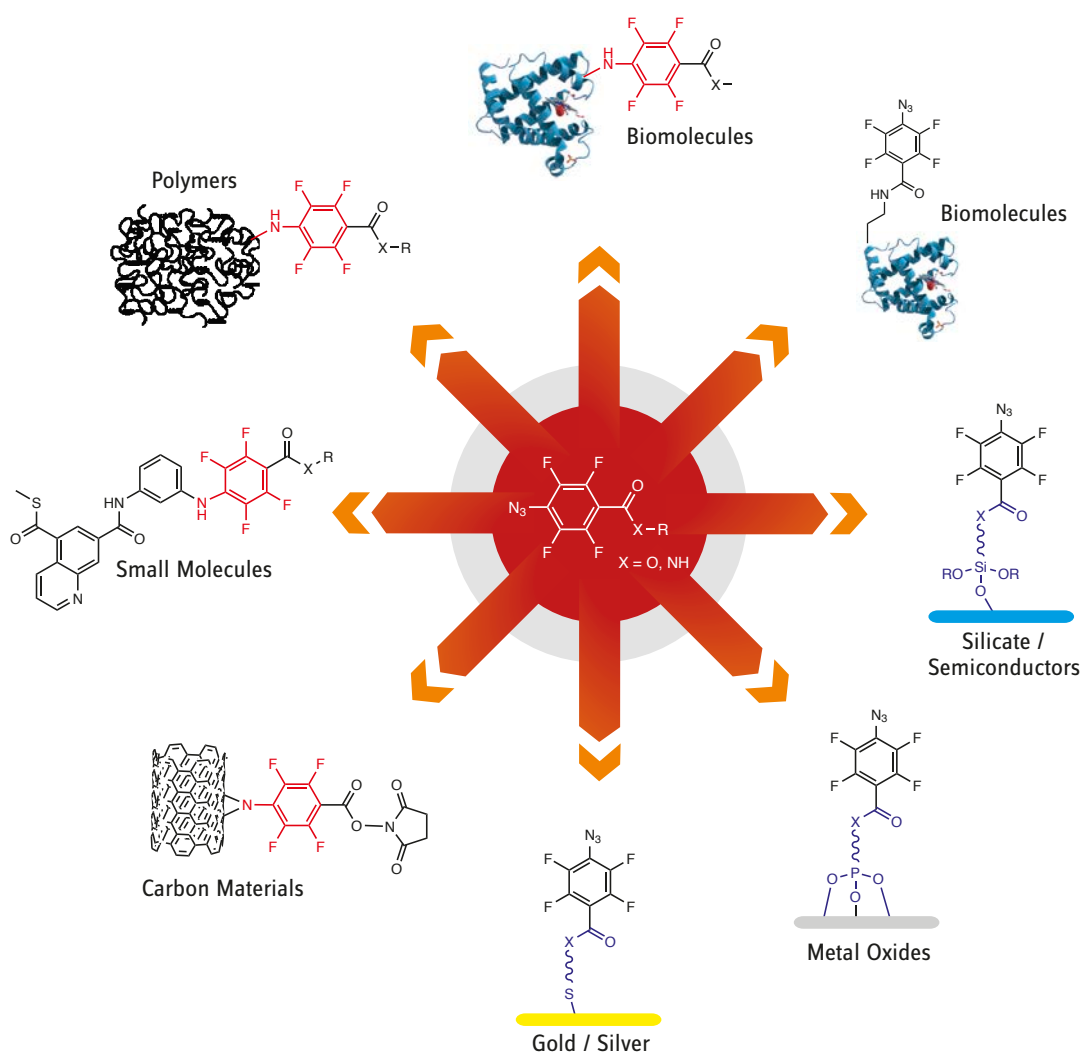
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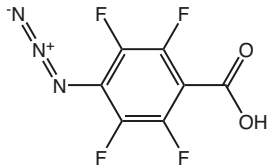
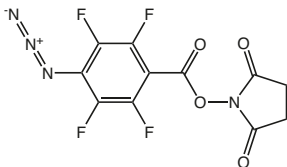
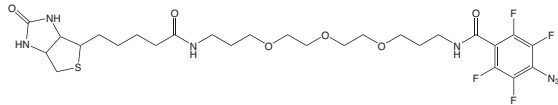
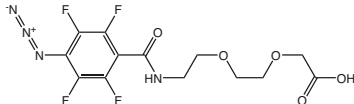
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2.2. Tetrafluorophenyl-Azide-based Photo-Crosslinkers

Tetrafluorophenyl-azides follow a principle similar to diazirines. Upon irradiation with UV light (ca. 260 nm), a highly stabilized nitrene is formed. Nitrenes are the nitrogen analogs of carbenes (isoelectronic) and react in a comparable fashion. In terms of crosslinking yield and duration of irradiation, they compare favorably to benzophenones. Moreover, the azido group can also

undergo classical copper-catalyzed azide-alkyne cycloadditions. Tetrafluorophenyl-azido crosslinkers are also available with a short PEG-spacer for increased solubility (PEG2065), and as Biotin-TEG-ATFBA (PEG5000) for applications such as surface functionalization with biotin, or the biotinylation of biomacromolecules.



		Article No.	Quantity	Price		
RL-2035	ATFB-C004		RL-2035.0250	250 mg	€ 100,00	
			RL-2035.0500	500 mg	€ 180,00	
RL-2035.0001	1 g		€ 280,00			
RL-2035.0005	5 g		€ 1.000,00			
4-Azido-2,3,5,6-tetrafluorobenzoic acid CAS NO: 122590-77-6 FORMULA: C ₇ H ₄ F ₄ N ₃ O ₂ MOLECULAR WEIGHT: 235,1 g/mol						
RL-2045	ATFB-NHS		RL-2045.0100	100 mg	€ 96,00	
			RL-2045.0250	250 mg	€ 160,00	
			RL-2045.0500	500 mg	€ 288,00	
			RL-2045.1000	1 g	€ 448,00	
			RL-2045.5000	5 g	€ 1.600,00	
N-Succinimidyl 4-azido-2,3,5,6-tetrafluorobenzoate CAS NO: 126695-58-7 FORMULA: C ₁₁ H ₄ F ₄ N ₄ O ₄ MOLECULAR WEIGHT: 332,17 g/mol						
PEG2065	Biotin-TEG-ATFB		PEG2065.0025	25 mg	€ 250,00	
			PEG2065.0100	100 mg	€ 425,00	
Biotin-triethylenglycol-(p-azido-tetrafluorobenzamide) CAS NO: 1264662-85-2 FORMULA: C ₂₇ H ₃₇ F ₄ N ₇ O ₆ S MOLECULAR WEIGHT: 663,68 g/mol						
PEG5000	ATFB-O₂Oc			PEG5000.0100	100 mg	€ 120,00
				PEG5000.0250	250 mg	€ 200,00
		PEG5000.0500		500 mg	€ 360,00	
		PEG5000.1000		1 g	€ 560,00	
		PEG5000.5000		5 g	€ 2.000,00	
{2-[2-(4-Azido-2,3,5,6-tetrafluorobenzoyl-amino)ethoxy]ethoxy}acetic acid FORMULA: C ₁₃ H ₁₂ F ₄ O ₅ MOLECULAR WEIGHT: 380,25 g/mol						

References:

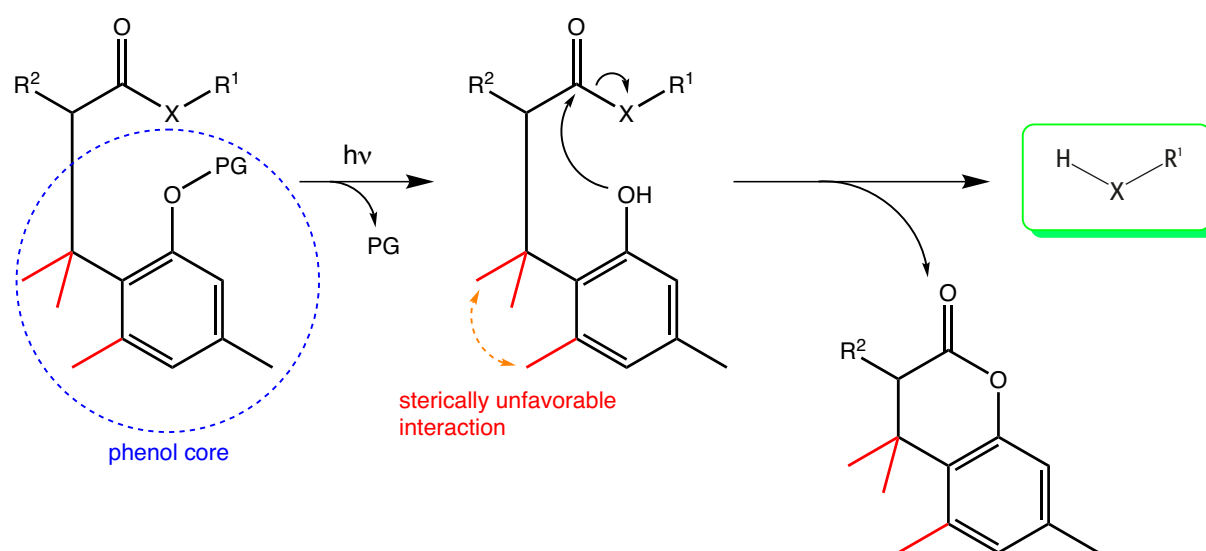
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3. Photoactivated Self-Cleaving Linkers and Protecting Groups via Trimethyl Lock

Iris Biotech introduces a series of self-immolative compounds that find application as protecting groups, linkers, or amino acid derivatives (Spr = stimulus-responsive peptide bond cleaving residue). The self-cleavage is induced by irradiation with UV light (ca. 350 - 365 nm) that leads to the unmasking of a hydroxyl group of a 2-alkyl-3,5-dimethyl phenol moiety. The photocleavable group is either *o*-nitrobenzyl or *o*-nitroveratryl, which can be cleaved at wavelengths > 350 nm. Since wavelengths above 350 nm tend to be unproblematic for biomacromolecules, this technique is especially interesting for cell-based systems.

The liberated OH-group serves as a nucleophile that intramolecularly cleaves ester (or amide) bonds at neutral

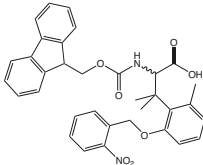
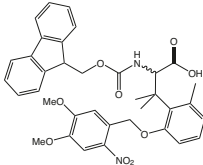
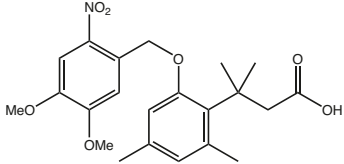
pH and room temperature by cyclization via a six-membered transition state. This reaction is greatly accelerated since the sterically unfavorable interaction between the methyl group at the 3-position of the phenol core and the two geminal CH₃-groups on the alkyl chain (in β-position to the ester or amide carbonyl group) favor conformations that bring the phenolic OH-group and the neighboring carbonyl function into closer vicinity. This phenomenon is termed the gem-dialkyl effect, for which a theory was first proposed by Thorpe and Ingold in 1915 ("Thorpe-Ingold effect"). An alternative explanation for this effect was posited by Bruice and Pandit in 1960 ("reactive rotamer effect").



Principle of photoactivated trimethyl lock; PG = o-nitrobenzyl or o-nitroveratryl; X = O or NH; R¹-XH = target molecule; R² = H or NH-alkyl.

The applications for this approach are numerous. Incorporation of a Spr-residue into a peptide sequence enables the photoactivated self-cleavage of the peptide at the position of said residue. This technique allows for the intracellular removal of e.g. cell penetrating peptides or localization sequences from a bioactive molecule. The Spr-residue can also be used as a photolabile

linker in order to reversibly connect a moiety such as biotin to a molecule of interest. Finally, while *o*-nitroveratryl itself is a valuable protecting group for sulfhydryl groups, the combination of *o*-Nv with a trimethyl lock moiety also allows for its use as a protecting group for hydroxyl and amino functions.

	Article No.	Quantity	Price
FAA7190 Fmoc-Spr(oNB)-OH N-alpha-(9-Fluorenylmethoxycarbonyl)-beta,beta-dimethyl-(2-methyl-6-(2-nitrobenzyl-oxyl)phenyl)alanine (rac.) CAS NO: 1032400-98-8 FORMULA: $C_{34}H_{32}N_2O_7$ MOLECULAR WEIGHT: 580,63 g/mol	FAA7190.0000	please inquire	
			
FAA7200 Fmoc-Spr(oNv)-OH N-alpha-(9-Fluorenylmethyl)-N-alpha-(9-Fluorenylmethoxy)carboxyl)-beta,beta-dimethyl-(2-methyl-6-(2-nitroveratryl)phenyl)alanine (rac.)xycarbonyl)-beta,beta-dimethyl-(2-methyl-6-(2-nitroveratryl)phenyl)alanine (rac.) CAS NO: 1228829-20-6 FORMULA: $C_{36}H_{36}N_2O_9$ MOLECULAR WEIGHT: 640,68 g/mol	FAA7200.0000	please inquire	
			
RL-2970 Photo-Trimethyl-Lock 3-(2-Nitroveratryl-4,6-dimethylphenyl)-3-methylbutyric acid CAS NO: 2095134-25-9 FORMULA: $C_{22}H_{27}NO_3$ MOLECULAR WEIGHT: 417,45 g/mol	RL-2970.0000	please inquire	
			

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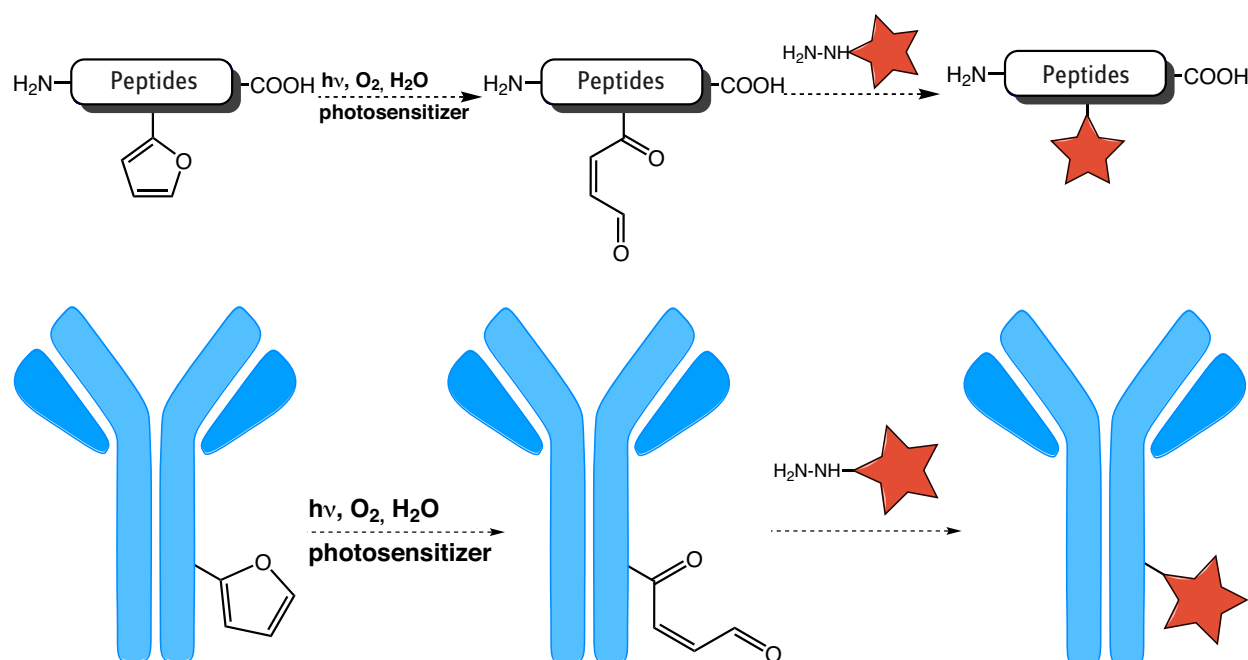


4. Furfuryl-Alanine for Side Chain Modification and Bioconjugation

2-Furyl-alanine can be incorporated into peptides *via* SPPS or by using enzymatic approaches. UV-irradiation in the presence of oxygen and a photosensitizer converts furyl-alanine to an intermediate that selectively reacts with certain nucleophiles. This property can be employed for site-specific labeling of peptides and proteins.

Labeling with different tags and reporter groups is a pivotal technique for the elucidation of peptide and protein function.

A novel and innovative approach is the site-specific labeling using the unnatural amino acid 2-furyl-alanine. UV-irradiation in the presence of oxygen and a photosensitizer converts furyl-alanine to an unsaturated dicarbonyl compound. This intermediate selectively reacts with certain nucleophiles such as hydrazine derivatives of dyes or fluorescent labels. This reaction can be used for the site-specific labeling of peptides and proteins and can be carried out in aqueous solution.



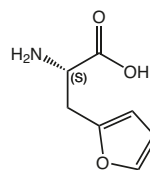
Site-specific labeling of peptides and proteins using 2-Furyl-alanine.

Iris Biotech offers Fmoc-L-Ala(2-Furyl)-OH suitable for SPPS, as well as H-L-Ala(2-Furyl)-OH which can be incorporated into proteins using the amber suppression methodology.

HAA2930 H-L-Ala(2-Furyl)-OH**3-(2-Furyl)-L-alanine**

CAS NO: 127682-08-0

MOLECULAR WEIGHT: 155,15 g/mol



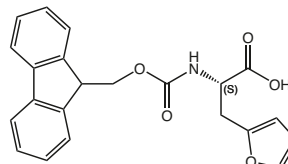
Article No.	Quantity	Price
HAA2930.0250	250 mg	€ 250,00
HAA2930.0001	1 g	€ 750,00
HAA2930.0005	5 g	€ 2.000,00

FAA4250 Fmoc-L-Ala(2-Furyl)-OH**N-alpha-(9-Fluorenylmethyloxycarbonyl)-3-(2-furyl)-L-alanine**

CAS NO: 159611-02-6

FORMULA: C₂₂H₁₉NO₅

MOLECULAR WEIGHT: 377,39 g/mol



FAA4250.0250	250 mg	€ 95,00
FAA4250.0001	1 g	€ 250,00
FAA4250.0005	5 g	€ 900,00

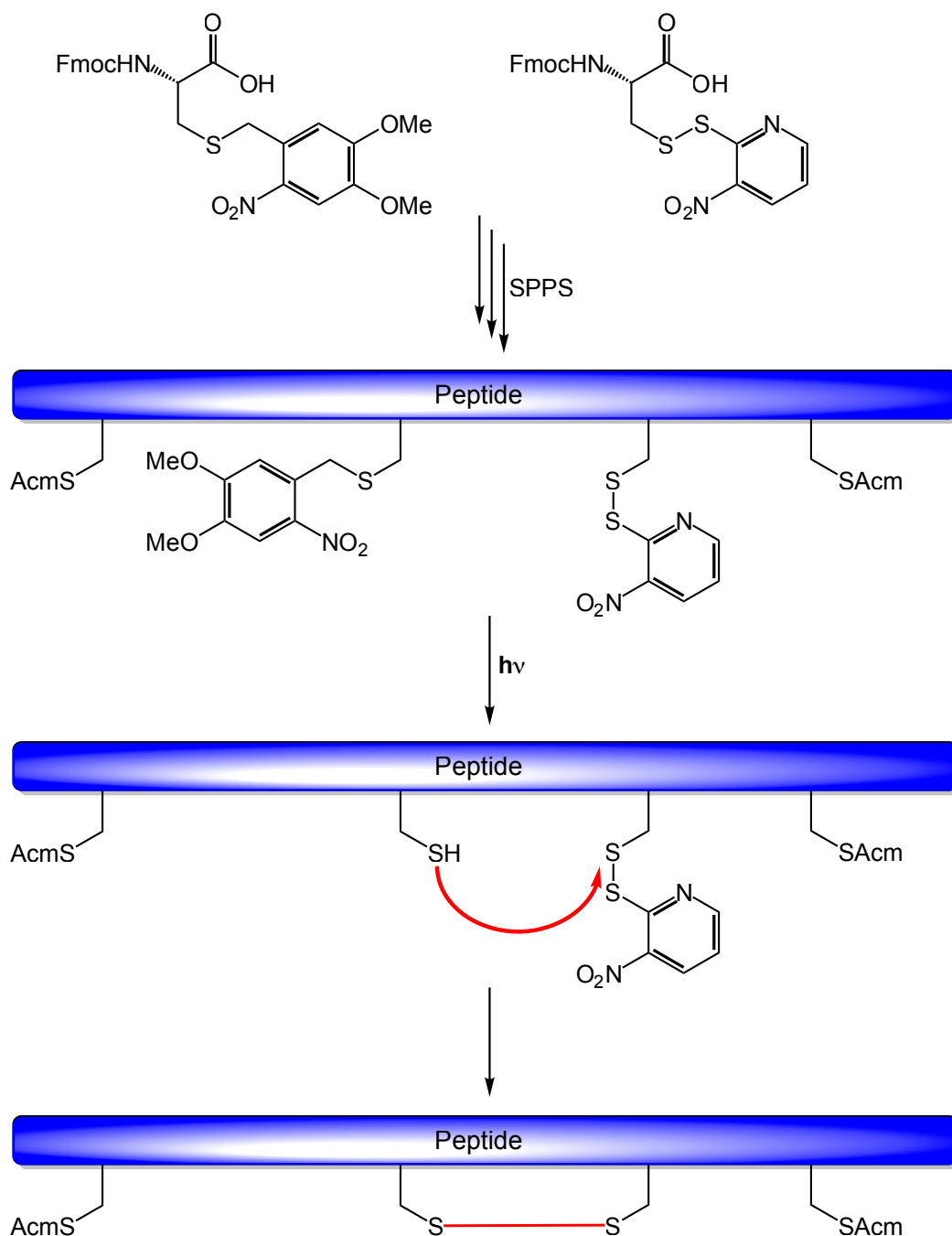
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5. *o*-Nitroveratryl Protected Cysteine for Disulfide Bridge Formation

2-Nitroveratryl (*o*Nv) is a photolabile orthogonal protecting group that is compatible with SPPS protocols and can be cleaved by irradiation with UV light (350 nm) under ambient conditions. Combination with S-pyridinesulfonyl activation allows for rapid

in situ disulfide bond formation. In order to demonstrate the versatility of this approach, it was applied to the synthesis of a number of model peptides: oxytocin, alpha-conotoxin Iml, and human insulin.



Selective deprotection of 2-nitroveratryl in the presence of other cysteine protecting groups, and subsequent disulfide bond formation with an Nps-functionalized cysteine.

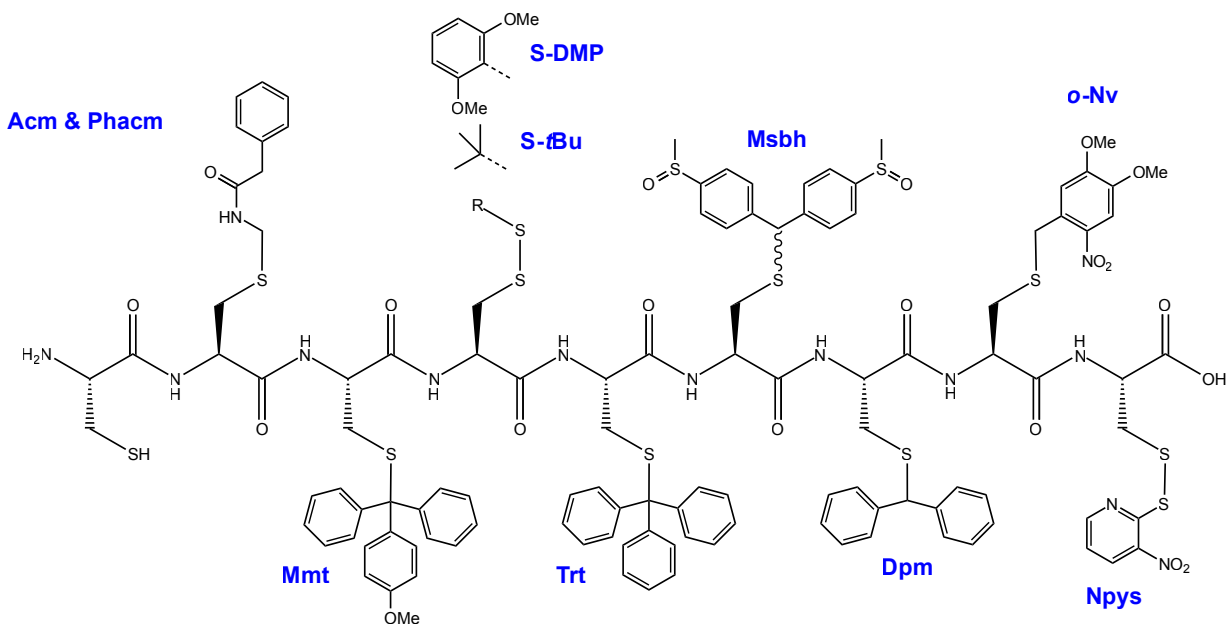
	Article No.	Quantity	Price
FAA3970 Fmoc-L-Cys(oNv)-OH N-alpha-(9-Fluorenylmethoxycarbonyl)-S-(2-nitroveratryl)-L-cysteine CAS NO: 214633-71-3 FORMULA: $C_{27}H_{26}N_2O_8S$ MOLECULAR WEIGHT: 538,57 g/mol	FAA3970.0250	250 mg	€ 475,00
	FAA3970.1000	1 g	€ 1.400,00
FAA1975 Fmoc-L-Cys(Npys)-OH N-alpha-(9-Fluorenylmethoxycarbonyl)-S-(3-nitro-2-pyridylthio)-L-cysteine FORMULA: $C_{23}H_{19}N_3O_6S_2$ MOLECULAR WEIGHT: 497,54 g/mol	FAA1975.0001	1 g	€ 275,00
	FAA1975.0005	5 g	€ 850,00

References:

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Is one set of orthogonal side chain protecting groups for Cysteine not enough?

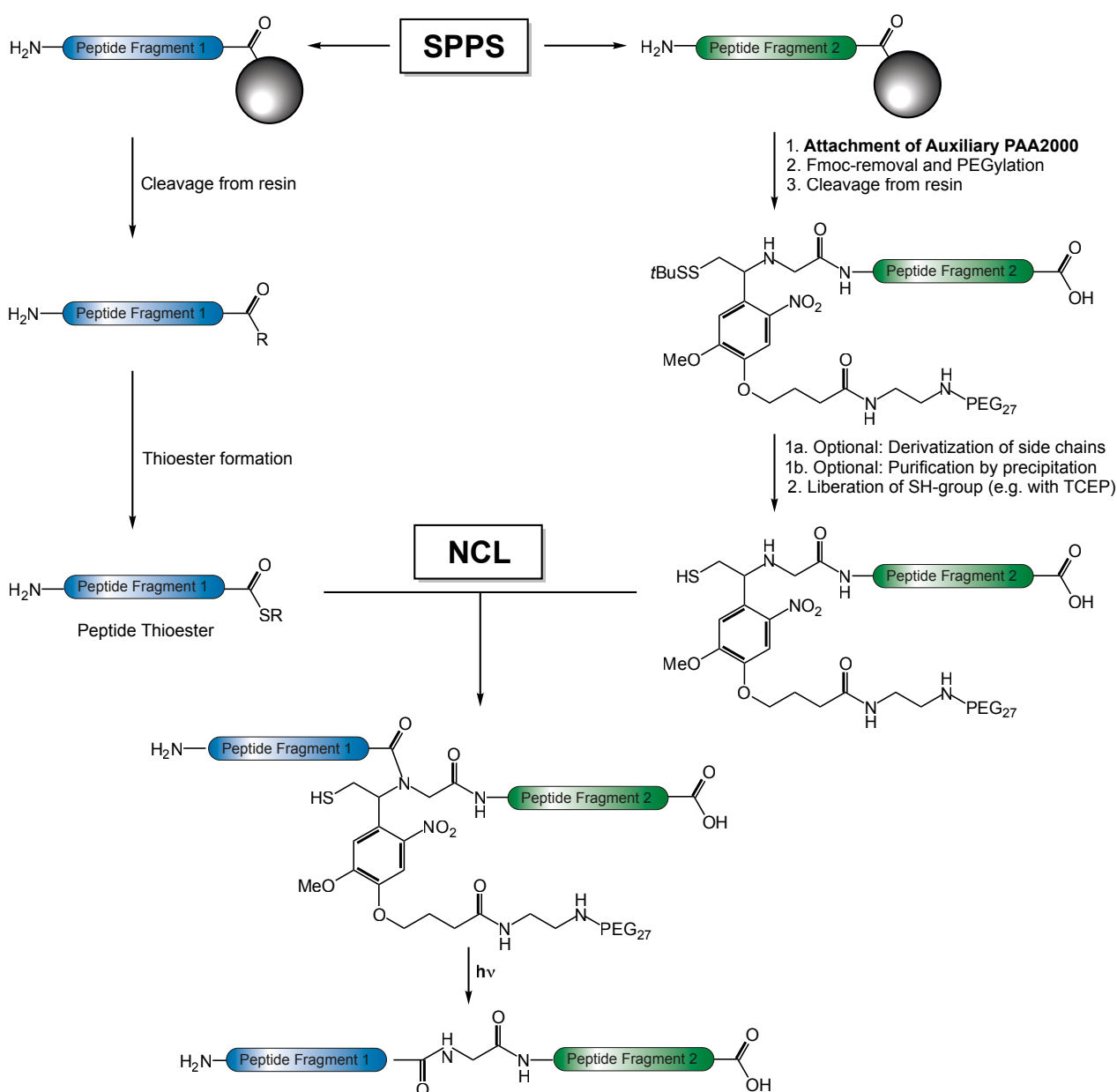
Iris Biotech offers a whole range of Cys protecting groups for disulfide bond formation:



6. Photocleavable Auxiliary Reagent for Native Chemical Ligation

Native Chemical Ligation is one of the most powerful tools for the preparation of complex peptides and small proteins. However, the classical variant of NCL requires an N-terminal cysteine at the ligation site. Iris Biotech presents an innovative auxiliary reagent for NCL that can be incorporated in place of a glycine residue.

Since glycine usually occurs several times in a peptide sequence, this approach significantly increases variability regarding the choice of possible ligation sites. In Native Chemical Ligation, the auxiliary's SH-group mimics the action of an N-terminal cysteine's sulfhydryl group.



Native Chemical Ligation utilizing the photocleavable NCL-auxiliary glycine building block (PAA2000).

Following SPPS, the auxiliary is attached to the N-terminus of a peptide sequence in lieu of a glycine residue. The auxiliary's Fmoc-protected amino functionality can subsequently be deprotected and functionalized, e.g. with a monodisperse PEG. PEGylation is useful for increasing the solubility of peptide fragments, and for facilitating their purification by precipitation with EtOH/Et₂O. These properties are especially valuable if the peptide's amino acid side chains are supposed to be further

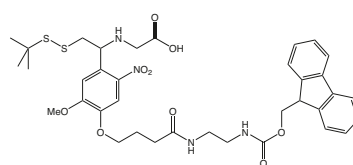
derivatized post-SPPS, for example by enzymatic glycosylation. Following NCL, the auxiliary can be conveniently removed by irradiation with UV-light (10 min in water or water/acetonitrile). This method is particularly useful for the synthesis of sophisticated peptides such as glycopeptides, where cost- and labor-intensive short sequences can be prepared separately, and subsequently conjugated to long fragments synthesized in a standard manner.

PAA2000 tBu-SS-Photo(Fmoc)-Gly-OH

Photocleavable-NCL-auxiliary-Gly-OH

FORMULA: C₃₆H₄₄N₄O₉S₂

MOLECULAR WEIGHT: 740,89 g/mol



Article No.	Quantity	Price
PAA2000.0025	25 mg	€ 300,00
PAA2000.0100	100 mg	€ 750,00

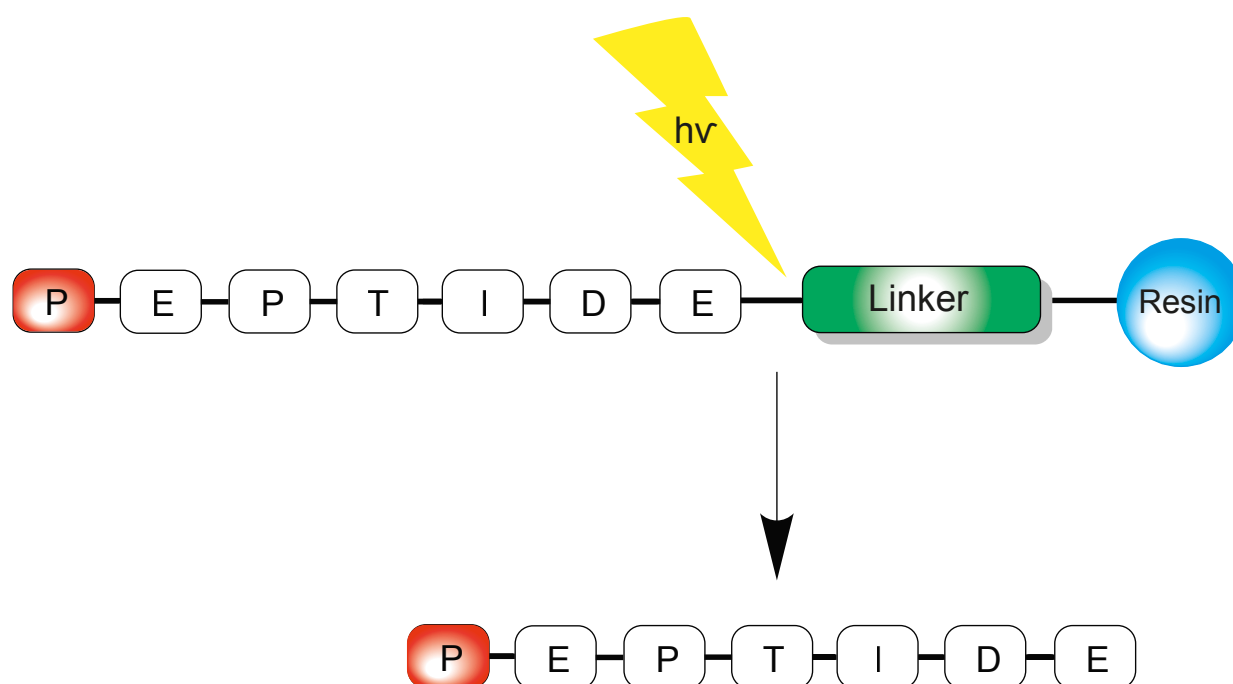
References:

- Ein PEGyliertes, lichtspaltbares Auxiliar für die sequenzielle enzymatische Glykosylierung und native chemische Ligation von Peptiden; Claudia Bello, Shuo Wang, Lu Meng, Kelly W. Moremen, Christian Becker; *Angew. Chem.* 2015; **127**: 7823-7828; DOI:10.1002/ange.201501517.
- A PEGylated Photocleavable Auxiliary Mediates the Sequential Enzymatic Glycosylation and Native Chemical Ligation of Peptides; Claudia Bello, Shuo Wang, Lu Meng, Kelly W. Moremen, Christian Becker; *Angew. Chem. Int. Ed.* 2015; **54**: 7711-7715; DOI:10.1002/anie.201501517.

Notes

7. Photo-Linker for Solid Phase Synthesis of Peptide Amides and Acids

Peptide linkers are usually cleaved under acidic conditions or using two-step procedures. Photocleavage proceeds under neutral conditions using UV light and can either be performed in batch or using flow chemistry.



Furthermore, photolabile linkers are orthogonal to standard peptide chemistry reaction conditions, thus enabling the use of a wide variety of amino acid protecting groups. Two different photolabile linkers are available for your convenience:

Photo-linker for the synthesis of C-terminal carboxylic acids

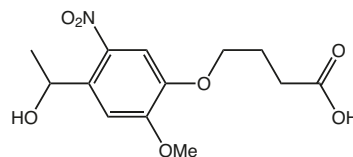
RL-2150 Acid-Photo-Linker

4-(4-(1-hydroxyethyl)-2-methoxy-5-nitrophenoxy)butanoic acid

CAS NO: 175281-76-2

FORMULA: C₁₃H₁₇NO₄

MOLECULAR WEIGHT: 299,28 g/mol



Article No.	Quantity	Price
RL-2150.0250	250 mg	€ 175,00
RL-2150.1000	1 g	€ 350,00
RL-2150.5000	5 g	€ 1.400,00

Fmoc-amino photo-linker for the synthesis of peptide amides

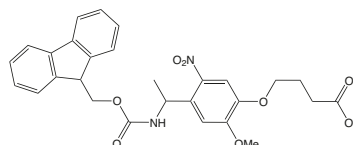
RL-1026 Fmoc-Photo-Linker

4-[4-[1-(9-Fluorenylmethyloxycarbonylamino)ethyl]-2-methoxy-5-nitrophenoxy]butanoic acid

CAS NO: 162827-98-7

FORMULA: C₂₈H₂₈N₂O₈

MOLECULAR WEIGHT: 520,56 g/mol

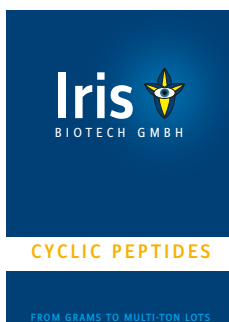


Article No.	Quantity	Price
RL-1026.0001	1 g	€ 250,00
RL-1026.0005	5 g	€ 900,00

References:

- Continuous Photochemical Cleavage of Linkers for Solid-Phase Synthesis; M. Hurevich, J. Kandasamy, B. M. Ponnappa, M. Collot, D. Kopetzki, D. T. McQuade and P. H. Seeberger; *Organic Letters* 2014; **16**: 1794-1797. doi:10.1021/ol500530q
- Photolytic Mass Laddering for Fast Characterization of Oligomers on Single Resin Beads; K. Burgess, C. I. Martinez, D. H. Russell, H. Shin and A. J. Zhang; *The Journal of Organic Chemistry* 1997; **62**: 5662-5663. doi:10.1021/jo970866w
- Direct Monitoring of Organic Reactions on Polymeric Supports; M. R. Carrasco, M. C. Fitzgerald, Y. Oda and S. B. H. Kent; *Tetrahedron Letters* 1997; **38**: 6331-6334. doi:http://dx.doi.org/10.1016/S0040-4039(97)01456-1

Available Booklets:

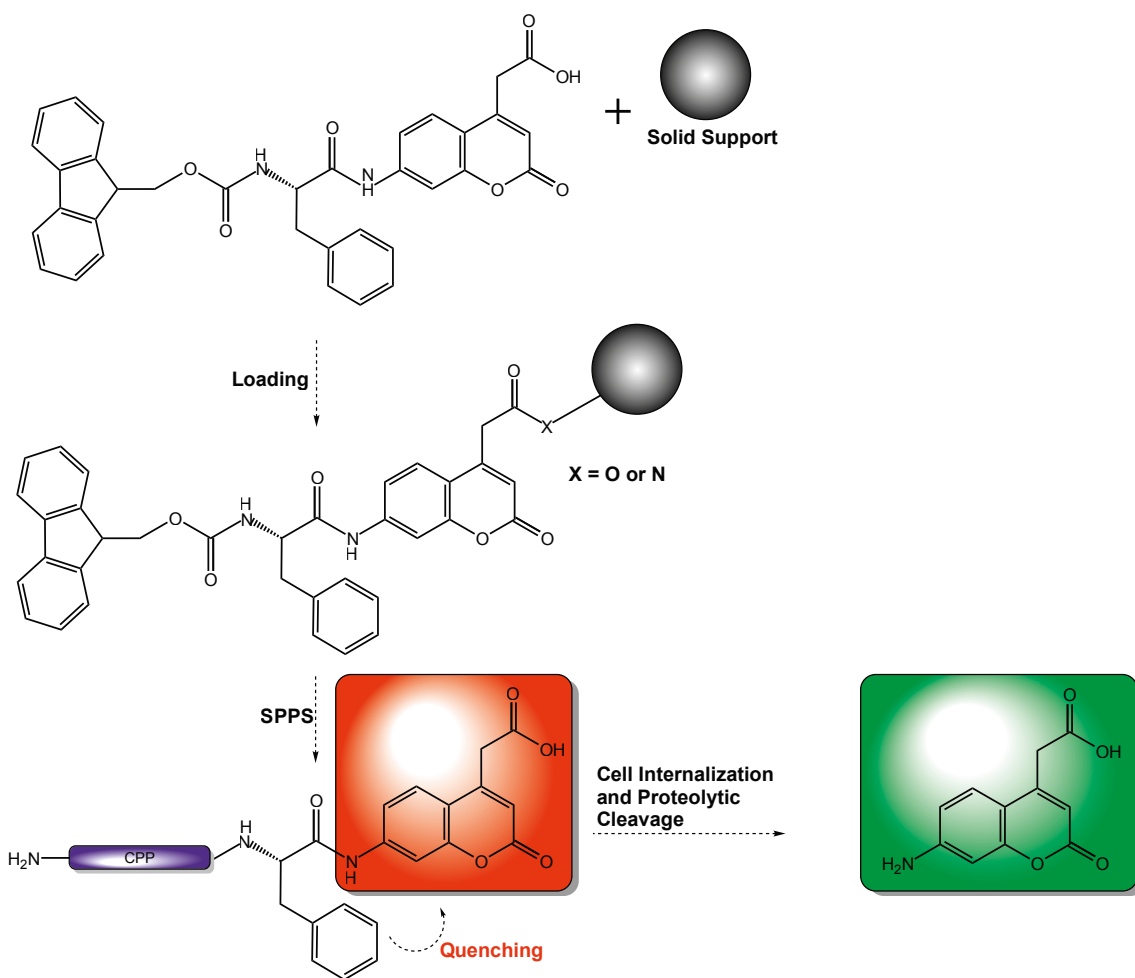
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Iris Booklets here:

8. Related Products

8.1. Fmoc-Phe-Aca: Fluorescent Internalization Reporter for Cell Penetrating Peptides (CPPs)

The unnatural amino acid Aca (7-amino-coumarin-4-acetic acid) is a coumarin derivative and thus exhibits fluorescence. When incorporated into a peptide C-terminally of phenylalanine, Aca is a useful reporter group for the successful internalization of CPPs. The phenyl moiety of Phe quenches the fluorescence of Aca. Internalization of the CPP containing Phe-Aca leads to proteolytic cleavage of the Phe-Aca bond and thus to fluorescence.

However, the peptide bond formation between Phe and Aca is considered to be a difficult coupling and often leads to low coupling yields. For your convenience, Iris Biotech offers Fmoc-Phe-Aca as building block suitable for SPPS. This pseudodipeptide can be coupled to the resin of your choice and subsequently elongated to prepare your target cell penetrating peptides.



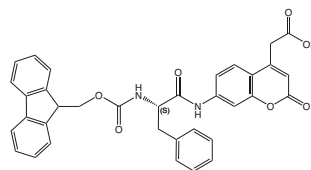
Loading of a solid support, subsequent elongation to CPPs and function as a reporter group for successful CPP internalization.

FDP1240 Fmoc-L-Phe-Aca-OH

7-[N-alpha-(9-Fluorenylmethyloxycarbonyl)-L-phenylalaninyl-amido]-coumarin-4-acetic acid

FORMULA: $C_{35}H_{28}N_2O_7$

MOLECULAR WEIGHT: 588,61 g/mol



Article No.	Quantity	Price
FDP1240.0100	100 mg	€ 195,00
FDP1240.0001	1 g	€ 1.250,00

References:

- Rapid and general profiling of protease specificity by using combinatorial fluorogenic substrate libraries; J. L. Harris, B. J. Backes, F. Leonetti, S. Mahrus, J. A. Ellman and C. S. Craik; *Proceedings of the National Academy of Sciences* 2000; **97**: 7754-7759. doi:10.1073/pnas.140132697
- Expedient Solid-Phase Synthesis of Fluorogenic Protease Substrates Using the 7-Amino-4-carbamoylmethylcoumarin (ACC) Fluorophore; D. J. Maly, F. Leonetti, B. J. Backes, D. S. Dauber, J. L. Harris, C. S. Craik and J. A. Ellman; *The Journal of Organic Chemistry* 2002; **67**: 910-915. doi:10.1021/jo016140o



Are you looking for different fluorescent reporter groups?

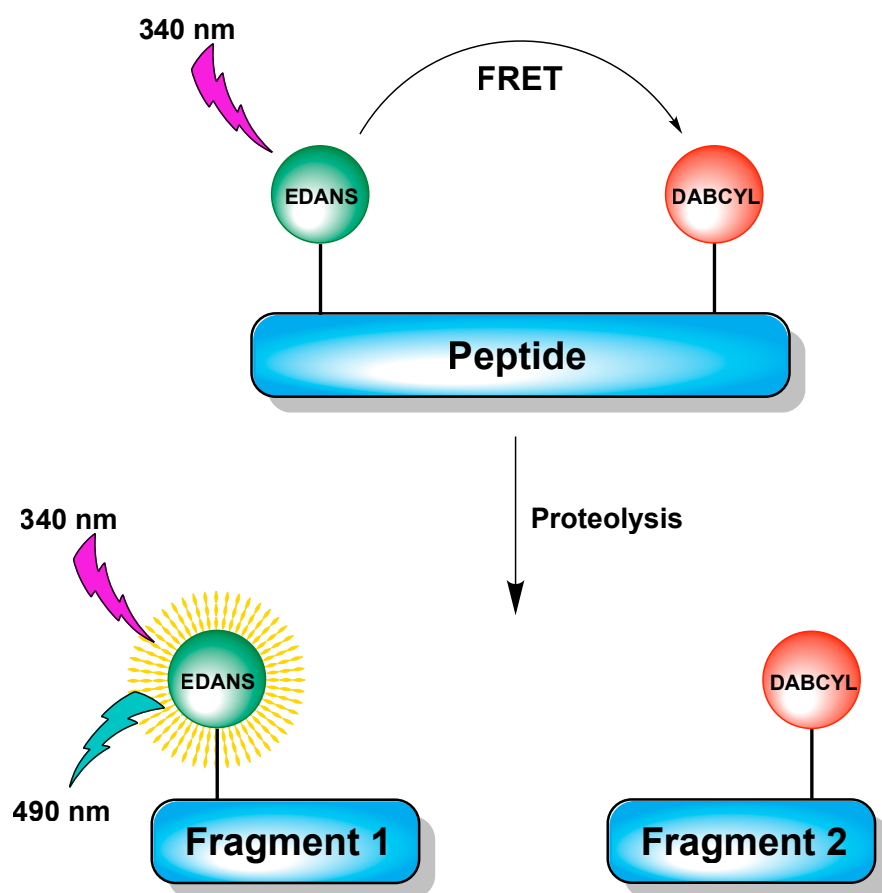
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Check out our Website for our new range of Indocyanine Green (ICG) dyes!

8.2. Building Blocks for the SPPS of FRET-based fluorogenic protease substrates

The classic FRET pair EDANS and DABCYL is now available linked to Fmoc amino acid building blocks that can be readily used in SPPS. Conveniently synthesize your own custom protease substrates!



Screening of protease activity using FRET-based fluorogenic protease substrates

Traditionally, fluorogenic protease substrates for the screening of protease activity are prepared by peptide synthesis and subsequent regioselective deprotection and functionalization with a fluorophore/quencher pair such as EDANS and DABSYL. For your convenience, we now offer a glutamate and a lysine building block functionalized with EDANS and DABSYL, respectively. Those building blocks are suitable for peptide synthesis using the Fmoc strategy and will allow you to prepare custom protease substrates without additional tedious and yield-reducing functionalization steps post-SPPS.

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